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(54) Title: COMPOSITIONS COMPRISING SILICONE-IN-WATER EMULSIONS AND FRAGRANCES AND HAIR CARE PREPARATIONS COMPRISING SUCH COMPOSITIONS

(57) Abstract: The present invention relates to compositions comprising silicone-in-water emulsions and fragrance compositions. It also relates to hair care preparations comprising such compositions, which provide improved release of fragrance when applied to the hair and improved deposition of the Iragrance on the hair. The invention is Iurther related to methods of cleansing hair, which comprises applying a hair care preparation to the hair.

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5 FRAGRANCES AND HAIR CARE PREPARATIONS COMPRISING SUCH COMPOSITIONS

The present application is based on U.S. provisional application 60/409,393, filed September 10, 2002, which is hereby incorporated herein by reference in its entirety.

FIELD OF THE INVENTION

The present invention relates to compositions comprising silicone-inwater emulsions and fragrance compositions. It also relates to hair care preparations comprising such compositions, which provide improved release of fragrance when applied to the hair and improved deposition of the fragrance on the hair. The invention is further related to methods of cleansing hair, which comprises applying a hair care preparation to the hair.

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BACKGROUND OF THE INVENTION

Silicone-in-water emulsions are well known for use in cosmetic preparations. Hair care preparations with silicone-in-water emulsions have been described, for example, in WO200203936, EP1093809, and WO200139735. Silicone is lubricious and acts to enhance the properties of the hair by improving the feel, the shine, and the ease of combing the hair. These compositions may also include fragrances, which are intended to be released during the application of the composition and, preferably, be deposited onto the hair for subsequent release. There is, however, still a need for hair care preparations that can provide improved deposition of fragrances or at least a deposition that will result in a longer lasting release of fragrance from the hair, without losing the favorable sensory and aesthetic performance qualities of the product.

It has now surprisingly been found that if an emulsion of a silicone polymer of certain minimum viscosity is introduced in combination with certain fragrances in the cosmetic preparation, better deposition of the fragrance can be achieved without any loss of the essential properties of hair care preparations. The better deposition particularly occurs upon joint introduction of the silicone polymer and the fragrance composition in a single emulsion composition.

SUMMARY OF THE INVENTION

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In a first aspect of the invention, there is provided a composition comprising i) a fragrance composition; and ii) a silicone-in-water emulsion comprising at least one surfactant, wherein the dispersed silicone has a viscosity of 100 million mm²s⁻¹ or more and a particle size of less than 2 μm. There is further provided a composition comprising: i) a fragrance composition, wherein the fragrance composition comprises a) at least 10.0% by weight of a fragrance composition of one or more perfume ingredients having a boiling point of 250 °C or less and ClogP of less than 3.0; b) at least 15% by weight of a fragrance composition of a one or more perfume ingredients having a boiling point of 250 °C or less and ClogP of 3.0 or more; and c) at least 20% by weight of a fragrance composition of one or more perfume ingredients have a boiling point of greater than 250 °C; and ii) a silicone-in-water emulsion comprising at least one surfactant.

The invention is directed, in a second aspect, to a composition, wherein the composition is a hair care preparation, comprising: i) a fragrance composition; and ii) a silicone-in-water emulsion comprising at least one surfactant, wherein the dispersed silicone has a viscosity of 100 million mm²s⁻¹ or more and a particle size of less than 2 µm. There is further provided a composition, wherein the composition is a

hair care preparation, comprising: i) a fragrance composition, wherein the fragrance composition comprises a) at least 10.0% by weight of a fragrance composition of one or more perfume ingredients having a boiling point of 250 °C or less and ClogP of less than 3.0; b) at least 15% by weight of a fragrance composition of a one or more perfume ingredients having a boiling point of 250 °C or less and ClogP of 3.0 or more; and c) at least 20% by weight of a fragrance composition of one or more perfumc ingredients have a boiling point of greater than 250 °C; and ii) a silicone-in-water emulsion comprising at least one surfactant.

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The invention is directed, in a third aspect, to a method of cleansing hair, 10 which comprises applying to the hair a hair care preparation comprising: i) a fragrance composition; and ii) a silicone-in-water emulsion comprising at least one surfactant, wherein the dispersed silicone has a viscosity of 100 million mm²s⁻¹ or more and a particle size of less than 2 µm. The invention is further directed to a method of cleansing hair, which comprises applying to the hair a hair care preparation comprising: i) a fragrance composition, wherein the fragrance composition comprises a) at least 15 10.0% by weight of a fragrance composition of one or more perfume ingredients having a boiling point of 250 °C or less and ClogP of less than 3.0; b) at least 15% by weight of a fragrance composition of one or more perfume ingredients having a boiling point of 250 °C or less and ClogP of 3.0 or more; and c) at least 20% by weight of a 20 fragrance composition of one or more perfume ingredients have a boiling point of greater than 250 °C; and ii) a silicone-in-water emulsion comprising at least one surfactant.

In certain aspects, the aforementioned fragrance compositions may comprise a) at least 10.0% by weight of a fragrance composition of one or more perfume ingredients having a boiling point of 250 °C or less and ClogP of less than 3.0;

b) at least 20% by weight of a fragrance composition of one or more perfume ingredients having a boiling point of 250 °C or less and ClogP of 3.0 or more; and c) at least 30% by weight of a fragrance composition of one or more perfume ingredients have a boiling point of greater than 250 °C.

5 DETAILED DESCRIPTION OF THE INVENTION

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In one aspect of the invention, there is provided a composition comprising i) a fragrance composition; and ii) a silicone-in-water emulsion comprising at least one surfactant, wherein the dispersed silicone has a viscosity of 100 million mm²s⁻¹ or more and a particle size of less than 2 µm. There is further provided a composition comprising: i) a fragrance composition, wherein the fragrance composition comprises a) at least 10.0% by weight of a fragrance composition of one or more perfume ingredients having a boiling point of 250 °C or less and ClogP of less than 3.0; b) at least 15% by weight of a fragrance composition of one or more perfume ingredients having a boiling point of 250 °C or less and ClogP of 3.0 or more; and c) at least 20% by weight of a fragrance composition of one or more perfume ingredients have a boiling point of greater than 250 °C; and ii) a silicone-in-water emulsion comprising at least one surfactant. There is further provided a composition comprising: i) a fragrance composition, wherein the fragrance composition comprises a) at least 10.0% by weight of a fragrance composition of one or more perfume ingredients having a boiling point of 250 °C or less and ClogP of less than 3.0; b) at least 20% by weight of a fragrance composition of one or more perfume ingredients having a boiling point of 250 °C or less and ClogP of 3.0 or more; and c) at least 30% by weight of a fragrance composition of one or more perfume ingredients have a boiling point of greater than 250 °C; and ii) a silicone-in-water emulsion comprising at least one surfactant.

Silicone-in-water emulsions

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The silicone-in-water emulsion used in the present invention may be prepared by emulsion polymerisation methods well-known in the art, such as those described in U.S. 5,998,537 and EP 0874017, incorporated herein by reference. In summary, this method comprises a) mixing materials comprising a composition containing at least one polysiloxane, at least one organosilicon material that reacts with the polysiloxane by a chain extension reaction, a metal containing catalyst for said chain extension reaction, at least one surfactant, and water to form a mixture, and b) emulsifying the mixture. This method provides a route to silicone-in-water emulsions having a wide range of physical characteristics, in particular emulsions wherein the dispersed silicone has a high viscosity and low droplet size.

Chain extension reactions with polysiloxanes are known in the art and can involve, for instance, the hydrosilylation reaction in which an Si-H reacts with an aliphatically unsaturated group in the presence of a platinum or rhodium-containing catalyst forming polysiloxanes having some Si-(C)_p-Si links (wherein p = 1-6), which polysiloxanes may also be referred to as polysiloxane polysilalkylene copolymers. Alternatively, the chain extension reaction can involve the reaction of an Si-OH group (e.g., a hydroxy-terminated polysiloxane) with an alkoxy group (e.g., alkoxysilanes, silicates, or alkoxysiloxanes) in the presence of a metal containing catalyst forming polysiloxane materials.

The polysiloxanes used in the chain extension reactions described above generally comprise a substantially linear polymer of the structure (I):

In this structure, each R independently represents a hydrocarbon group having up to 20 carbon atoms, preferably 1 to 6 carbon atoms, such as an alkyl group (e.g., methyl, ethyl, propyl or butyl), an aryl group (e.g., phenyl), or the group required for the chain extension reaction described above ("reactive group", e.g., silicon-bonded hydrogen atoms, aliphatically unsaturated groups such as vinyl, allyl or hexenyl, hydroxy, alkoxy such as methoxy, ethoxy or propoxy, alkoxy-alkoxy, acetoxy, amino and the like), provided that on average there is from one to two reactive groups (inclusive) per polymer, and n is a positive integer greater than one. Preferably, a majority, more preferably >90%, and most preferably >98% of the reactive groups are substituted on the terminal silicone atom in the siloxane. Preferably, n is an integer which results in polysiloxanes with viscosities between about 1 and about 1 million mm²s⁻¹, preferably from 1000 to 100,000 mm²s⁻¹.

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The polysiloxane (I) can have a small amount of branching (e.g., less than 2 mole % of the siloxane units), but preferably, wherein the polymers are substantially linear, more preferably completely linear. Moreover, the R groups can be substituted with, for example, nitrogen containing groups (e.g., amino groups), epoxy groups, sulfur-containing groups, silicon-containing groups, oxygen-containing groups and the like. Preferably, however, at least 80% of the R groups are alkyl and, more preferably, the alkyl groups are methyl groups.

The organosilicon material that reacts with the polysiloxane by a chain extension reaction can be either a second polysiloxane or a material which acts as a

chain extension agent. If the organosilicon material is a polysiloxane, then it too will generally have the structure described above (I). In such a situation, one polysiloxane in the reaction will comprise one reactive group and the second polysiloxane will comprise a second reactive group that reacts with the first.

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If the organosilicon material comprises a chain extension agent, it can be a material such as a silane, a siloxane (e.g., disiloxane or trisiloxane) or a silazane. For example, a composition comprising a polysiloxane according to the above structure (I) which has at least one Si-OH group can be chain extended by using an alkoxysilane (e.g., a dialkoxysilane or trialkoxysilane) in the presence of a tin or titanium containing catalyst.

The metal containing catalysts used in the above described chain extension reactions are often specific to the particular reaction. Such catalysts, however, are known in the art. Generally, they are materials containing metals such as platinum, rhodium, tin, titanium, copper, lead, etc.

In a preferred chain-extension reaction, a polysiloxane having at least one aliphatically unsaturated group, preferably an end-group, is reacted with an organosilicon material that is a siloxane or a polysiloxane having at least one Si-H group, preferably an end group, in the presence of a hydrosilylation catalyst. The polysiloxane having at least one aliphatically unsaturated group has the structure (I) wherein R and n are as defined above and provided that on average between one and two (inclusive) R groups comprise an aliphatically unsaturated group per polymer. Representative aliphatically unsaturated groups include vinyl, allyl, hexenyl and cyclohexenyl or a group R²CH=CHR³, where R² represents a divalent aliphatic chain linked to the silicon atom and R³ represents a hydrogen atom or an alkyl group. The organosilicon material having at least one Si-H group preferably has the above structure

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(I) wherein R and n are as defined above and provided that on average between one and two (inclusive) R groups comprise hydrogen atoms and n is 0 or a positive integer. This material can be a polymer or a lower molecular weight material such as a siloxane (e.g. a disiloxane or a trisiloxane).

The polysiloxane having at least one aliphatically unsaturated group and the organosilicon material having at least one Si-H group react in the presence of a hydrosilylation catalyst. Such catalysts are known in the art and include, for example, platinum and rhodium-containing materials. These catalysts may take any of the known forms, such as platinum or rhodium deposited on carriers such as silica gel or powdered charcoal, or other appropriate compounds such as platinic chloride, salts of platinum and chloroplatinic acids. A preferred material is chloroplatinic acid either as the commonly obtainable hexahydrate or the anhydrous form because of its easy dispersibility in organosilicon systems and its non-effect on colour of the mixture. Platinum or rhodium complexes may also be used, for example those prepared from chloroplatinic acid hexahydrate and divinyltetramethyldisiloxane. Generally, these catalysts are used in amounts of between about 0.0001 and 10 wt. % based on the weight of the composition (I).

In a second preferred chain extension reaction, a polysiloxane having at least one Si-OH group, preferably an end group, is reacted with an organosilicon material having at least one alkoxy group, preferably a siloxane having at least one Si-OR group, or an alkoxysilane having at least two alkoxy groups in the presence of a metal containing catalyst. In this case, the polysiloxane having at least one SiOH group has the structure (I) wherein R and n are as defined above and on average between one and two (inclusive) R groups comprise a hydroxyl group (OH). The organosilicon material having at least one alkoxy group can have the structure (I) wherein R and n are

as defined above and on average between one and two (inclusive) R groups comprise alkoxy groups, e.g., of the structure (OR) in which R is as defined above and n is 0 or a positive integer. Alternatively, the organosilicon material can be a silane of the structure $R_mSi(OR)_{4-m}$, wherein R is as defined above and m is 0 to 2. Other materials containing the alkoxy group (e.g., alkoxy-alkoxy) may also be used.

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A variety of metal catalysts for the reaction of an Si-OH group with an Si-OR group are known in the art and may be employed including, for example, organic metal compounds such as organotin salts, titanates, or titanium chelates or complexes. Examples include stannous octoate, dibutyltin dilaurate, dibutyltin diacetate, dimethyltin dineodecanoate, dibutyltin dimethoxide, isobutyl tin triceroate, dimethyltin dibutyrate, dimethyltin dineodecanoate, triethyltin tartrate, tin oleate, tin naphthenate, tin butyrate, tin acetate, tin benzoate, tin sebacate, tin succinate, tetrabutyl titanate, tetraisopropyl titante, tetraphenyl titante, tetraoctadecyl titanate, titanium naphthanate, ethyltriethanolamine titante, titanium diiso-propyl diethyl acetoacetate, titanium diisopropoxy diacetyl acetonate, and titanium tetra alkoxides where the alkoxide is butoxy or propoxy. Generally, these catalysts are used in amounts of between about 0.001 and 10 wt. % based on the weight of the composition (I).

Any composition containing at least one polysiloxane, at least one organosilicon material which reacts with said polysiloxane by a chain extension reaction, and a metal containing catalyst for said chain extension reaction can be used for the chain extension reaction.

The dispersed silicone in the emulsion used in the present invention has a viscosity of preferably 100 million mm 2 s $^{-1}$ or more, more preferably 150 million mm 2 s $^{-1}$ or more, and most preferably 170 million mm 2 s $^{-1}$ or more, and a particle size of preferably 2 μ m ore less, more preferably 1 μ m or less, and most preferably 0.5 μ m or

less. (Viscosities are measured at 25°C and atmospheric pressure.) It is also a preferred option that the particle size is small enough to cause the emulsion to be a microemulsion.

The emulsion preferably comprises at least 50% wt. silicone, more preferably at least 60 wt. %, even more preferably 75 wt. %. In addition to the polymerised silicone, the emulsion may also contain up to 2.5% by weight, based on the total weight of silicone, of a volatile silicone, which is mainly selected from cyclic siloxanes, preferably dimethyl cyclo polysiloxanes, and short chain polysiloxanes, preferably polydimethyl siloxanes with a chain length of no more than 10 siloxane units.

The preferred silicone-in-water emulsions of the invention are non-ionic and the silicone is a linear polysiloxane polysilalkylene copolymer, wherein the monovalent silicone substituents are substantially all C_{1-6} hydrocarbon groups and the copolymer has a viscosity of 150 million mm²s⁻¹ or more and a particle size of 1 μ m or less. In a more preferred embodiment, the silicone-in-water emulsion is a non-ionic emulsion comprising about 60 % by weight of a linear silicone copolymer formed by the hydrosilylation of a Si-H end-blocked polydimethyl siloxane and a vinyl-end-blocked polydimethyl siloxane, wherein the copolymer has a viscosity of about 170 mm²s⁻¹ and a particle size of about 0.5 μ m, and about 5 % by weight of non-ionic surfactants.

Surfactants

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The mixture used to form the emulsion also contains at least one surfactant. This can be a non-ionic surfactant, a cationic surfactant, an anionic surfactant, alkylpolysaccharides, amphoteric surfactants and the like.

Examples of non-ionic surfactants include polyoxyalkylene alkyl ethers, polyoxyalkylene sorbitan alkyl esters, polyoxyalkylene alkyl esters, and polyoxyalkylene alkylphenol ethers, polyethylene glycols, polypropylene glycols, and diethylene glycols. Particular examples include BRIJ35 and BRIJ30 (ICI Chemicals).

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Examples of cationic surfactants include quaternary ammonium hydroxides such as tetramethylammonium hydroxide, octyltrimethylammonium hydroxide, dodecyl-trimethyl ammonium hydroxide, hexadecyltrimethyl ammonium hydroxide, octyldimethylbenzylanımonium hydroxide, decyldimethylbenzyl ammonium hydroxide, didodecyldimethyl ammonium hydroxide, dioctadecyl dimethylammonium hydroxide, tallow trimethylammonium hydroxide and cocotrimethylammonium hydroxide as well as corresponding salts of these materials, fatty acid amines and amides and their derivatives and the salts of the fatty acid amines and amides including aliphatic fatty amines and their derivatives, homologs of aromatic amines having fatty chains, fatty amides derived from aliphatic diamines, fatty amides derived from disubstituted amines, derivatives of ethylene diamine, amide derivatives of amino alcohols, amine salts of long chain fatty acids, quaternary ammonium bases derived from fatty amides of disubstituted diamines quaternary ammonium bases of benzimidazolines, basic compounds of pyridinium and its derivatives, sulfonium compounds, quaternary ammonium compounds of betaine, urethanes of ethylene diamine, polyethylene diamines and polypropanolpolyethanol amines.

Examples of suitable anionic surfactants include alkyl sulfates such as lauryl sulfate, polymers such as acrylates/C₁₀₋₃₀ alkyl acrylate crosspolymer alkylbenzenesulfonic acids and salts such as hexylbenzenesulfonic acid, octylbenzenesulfonic acid, decylbenzenesulfonic acid, dodecylbenzenesulfonic acid, cetylbenzenesulfonic acid and myristylbenzenesulfonic acid; the sulfate esters of

monoalkyl polyoxyethylene ethers; alkylnapthylsulfonic acid; alkali metal sulforecinates, sulfonated glyceryl esters of fatty acids such as sulfonated monoglycerides of coconut oil acids, salts of sulfonated monovalent alcohol esters, amides of amino sulfonic acids, sulfonated products of fatty acid nitriles, sulfonated aromatic hydrocarbons, condensation products of naphthalene sulfonic acids with formaldehyde, sodium octahydroanthracene sulfonate, alkali metal alkyl sulfates, ester sulfates, and alkarylsulfonates.

Examples of suitable alkylpolysaccharides include, for example, materials of the structure R¹-O-(R²O)x-(G)y wherein R¹ represents a linear or branched alkyl group, a linear or branched alkenyl group or an alkylphenyl group, R² represent an alkylene group, G represents a reduced sugar, x denotes 0 or a positive integer and y denotes a positive integer as described, for example, in US Patent 5,035,832, incorporated herein by reference.

Examples of suitable amphoteric surfactants include cocamidopropyl betaine and cocamidopropyl hydroxysulfate.

The above surfactants may be used individually or in combination.

Fragrance compositions

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The fragrances of the present invention comprise a) at least 10.0% by

weight of the fragrance composition of one or more perfume ingredients having a

boiling point of 250 °C or less and ClogP of less than 3.0; b) at least 15% by weight of
the fragrance composition of one or more perfume ingredients having a boiling point
of 250 °C or less and ClogP of 3.0 or more, and c) at least 20% by weight of the
fragrance composition of one or more perfume ingredients have a boiling point of
greater than 250 °C. Examples of these three classes of perfume ingredients are

illustrated in Tables 1-3. The specific ingredients in these Tables are illustrative only and other materials familiar to those skilled in the art that fall within the guidelines above can also be incorporated into the fragrance compositions of the present invention.

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The fragrance compositions of the invention preferably comprise at least 20.0% by weight of the fragrance composition of one or more perfume ingredients having a boiling point of 250 °C or less and ClogP of less than 3.0; at least 15% by weight of the fragrance composition of a one or more perfume ingredients having a boiling point of 250 °C or less and ClogP of 3.0 or more; and at least 30% by weight of the fragrance composition of one or more perfume ingredients have a boiling point of greater than 250 °C. The fragrances compositions of the invention more preferably comprise: a) 30.0 to 40.0 % by weight of the fragrance composition of one or more perfume ingredients having a boiling point of 250 °C or less and ClogP of less than 3.0; b) 20.0 to 25.0 % by weight of the fragrance composition of one or more perfume ingredients having a boiling point of 250 °C or less and ClogP of 3.0 or more; and c) 40.0 to 45.0 % by weight of the fragrance composition of one or more perfume ingredients have a boiling point of greater than 250 °C.

The fragrance compositions of the invention include as examples, but are not limited to fragrance 890656 and fragrance 862883. Fragrance 890656 comprises a) 31.03 % by weight of the fragrance composition of one or more perfume ingredients having a boiling point of 250 °C or less and ClogP of less than 3.0; b) 20.17 % by weight of the fragrance composition of a one or more perfume ingredients having a boiling point of 250 °C or less and ClogP of 3.0 or more; and c) 44.91 % by weight of the fragrance composition of one or more perfume ingredients have a boiling point of greater than 250 °C. Fragrance 890656 comprises 12.79 % Di propylene

glycol; 12.00 % Linalyl acetate; 9.00 % Hexyl cinnamic aldehyde; 7.50 % Tonalide; 7.38 % Benzyl salicylate; 6.05% Phenyl ethyl alcohol; 4.89 % Linalool; 4.50 % Vertenex; 10.01 % apple base; 4.93 % ozonic base; 9.84 % woody base; 10.61 % floral base; and .50 % oils or resins. Apple base, ozonic base, woody base, and floral base are fragrances comprising one or more perfume ingredients that have the combined indicated characteristic. Fragrance 890656 has the fragrance description of citrus, green, lemon, melon, apple, woody, jasmin, herbal, peach, muguet, and musk.

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Fragrance 862883 comprises a) 34.08 % by weight of the fragrance composition of one or more perfume ingredients having a boiling point of 250 °C or less and ClogP of less than 3.0; b) 23.10 % by weight of the fragrance composition of 10 a one or more perfume ingredients having a boiling point of 250 °C or less and ClogP of 3.0 or more; and c) 42.46 % by weight of the fragrance composition of one or more perfume ingredients have a boiling point of greater than 250 °C. Fragrance 862883 comprises 18.98 % Di propylene glycol; 7.35 Verdox; 4.48 % Hedion; 4.47 % Linalool; 4.30 % Benzyl salicylate; 4.23 % Methyl hydrocinnamic aldehyde; 4.06 % 15 Dihydro myrcenol; 4.25 % Citrus oil; 3.56 % Benzyl acetate; 3.27 % Hexyl cinnamic aldehyde; 2.04 % Benzyl Benzoate; 1.36 % Linalyl acetate; 18.12 % apple base; 5.07 % floral sandelwood base; 4.91 % fruit base; 5.77 % green base; and 3.78 % oils or resins. Apple base, floral sandelwood base, fruit base, and green base are fragrances comprising one or more perfume ingredients that have the combined indicated 20 characteristic. Fragrance 862883 has the fragrance description of apple blossom, fruity, apple, floral, sandalwood, and musk.

The boiling points (BP) of many perfume ingredients, at standard 760 mmHg are given in, e.g., "Perfume and Flavor Chemicals (Aroma Chemicals)," Steffen Arctander, published by the author, 1969, incorporated herein by reference. The

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octanol/water partition coefficient of a fragrance ingredient is the ratio between its equilibrium concentrations in octanol and in water. The partition coefficients of the preferred perfume ingredients of the present invention may be more conveniently given in the form of their logarithm to the base 10, logP. The logP values of many perfume ingredients have been reported; for example, the Pomona92 database, available from Daylight Chemical Information Systems, Inc. (Daylight CIS), Irvine, Calif., contains many, along with citations to the original literature. However, the logP values are most conveniently calculated by the "CLOGP" program, also available from Daylight CIS. This program also lists experimental logP values when they are available in the Pomona database. The "calculated logP" (ClogP) is determined by the fragment 10 approach of Hansch and Leo (cf., A. Leo, in Comprehensive Medicinal Chemistry, Vol. 4, C. Hansch, P. G. Sammens, J. B. Taylor and C. A. Ramsden, Eds., p. 295, Pergamon Press, 1990, incorporated herein by reference). The fragment approach is based on the chemical structure of each perfume ingredient, and takes into account the numbers and types of atoms, the atom connectivity, and chemical bonding. The ClogP values, which 15 represent the octanol/water partition coefficient of a fragrance ingredient, are the most reliable and widely used estimates for this physicochemical property and are preferably used instead of the experimental logP values in the selection of perfume ingredients which are useful in the present invention.

Perfume materials with boiling point less than or equal to 250 °C and TABLE 1. than 3.0 CLogP less

	Name	BP	CLogP
25	Allyl Caproate	185	2.772
	Amyl Acetate	142	2.258
	Amyl Propionate	161	2.657
	Anisic Aldehyde	248	1.779
	Anisole	154	2.061
30	Benzaldehyde	179	1.480
_ 0	Benzyl Acetate	215	1.960

Benzyl Formate 202		Benzyl Acetone	235	1.739
Benzyl Formate 202 1.414 Benzyl Iso Valerate 246 2.887 Eagl Propionate 222 2.489 Beta Gamma Hexenol 157 1.337 Camphor Gum 208 2.117 laevo-Carvone 231 2.010 laevo-Carvone 230 2.203 Cinnamyl Formate 250 1.908 cis-Jasmone 248 2.712 cis-J-Hexenyl Acetate 169 2.243 Cuminic alcohol 248 2.531 Cuminic alcohol 248 2.531 Ciminic alcohol 248 2.531 Cuminic alcohol 248 2.531 Dimethyl Benzyl Carbinol 215 1.891 Dimethyl Benzyl Carbinyl Acetate 250 2.797 Ethyl Acetate 250 2.797 Ethyl Acetate 250 2.797 Ethyl Butyrate 181 0.333 Ethyl Butyrate 181 0.333 Ethyl Butyrate 121 1.729 Ethyl Hexyl Ketone 190 2.916 Ethyl Phenyl Acetate 212 2.640 Ethyl Phenyl Acetate 229 2.489 Eucalyptol 176 2.756 Fenchyl Alcohol 200 2.579 For Acetate (tricyclo Decenyl Acetate 175 2.357 Frutene (tricyclo Decenyl Propionate 200 2.260 Geraniol Hexenol 159 1.397 Hexenyl Acetate 172 2.787 Frutene (tricyclo Decenyl Propionate 200 2.260 Geraniol 159 1.397 Hexenyl Acetate 172 2.787 Hexyl Formate 155 2.381 Hydratropic Alcohol 219 1.582 Hydroxycitronellal 241 1.541 Isoamyl Alcohol 132 1.222 40 Isomenthone 210 2.831 Isoquinoline 243 2.080 Methyl Formate 249 2.00 Isoquinoline 243 2.080 Methyl Formate 202 2.929 Menthone 207 2.650 Methyl Acetophenone 208 2080 Methyl Amyl Ketone 152 2.886 Methyl Amyl Ketone 228 2.080 Methyl Amyl Ketone 252 2.080 Methyl Amyl Ketone 25		-	205	1.100
Benzyl Iso Valerate 246 2.887		•	202	1.414
5 Benzyl Propionate 222 2.489 Beta Gamma Hexenol 157 1.337 Camphor Gum 208 2.117 laevo-Carveol 227 2.265 d-Carvone 231 2.010 10 laevo-Carvene 230 2.203 Cinnamyl Formate 250 1.908 cis-Jasmone 248 2.712 cis-Jasmone 248 2.712 cis-Jakone 248 2.712 Cuminic alcohol 248 2.531 15 Cuminic aldehyde 236 2.780 Cyclal C 180 2.301 Dimethyl Benzyl Carbinol 215 1.891 Dimethyl Benzyl Carbinol 215 2.797 Ethyl Acetate 181 0.333 Ethyl Acetate		•	246	2.887
Beta Gamma Hexenol 157	5	•	222	2.489
laevo-Carveol			157	1.337
laevo-Carveol d-Carvone		Camphor Gum	208	2.117
d-Carvone		-	227	2.265
Cimamyl Formate 250 1.908 cis-Jasmone 248 2.712 cis-Jasmone 248 2.712 cis-Jasmone 248 2.712 cis-Jasmone 248 2.712 cis-Jasmone 248 2.713 15 Cuminic alcehol 248 2.531 15 Cuminic aldehyde 236 2.780 Cyclal C 180 2.301 Dimethyl Benzyl Carbinol 215 1.891 Dimethyl Benzyl Carbinyl Acetate 250 2.797 Ethyl Acetate 77 0.730 2.797 Ethyl Acetate 77 0.730 2.797 Ethyl Aceto Acetate 181 0.333 Ethyl Amyl Ketone 167 2.307 Ethyl Benzoate 212 2.640 Ethyl Butyrate 121 1.729 Ethyl Hexyl Ketone 190 2.916 2			231	2.010
Cimnamyl Formate 250 1.908 cis-Jasmone 248 2.712 cis-3-Hexenyl Acetate 169 2.243 Cuminic alcohol 248 2.531 15 Cuminic aldehyde 236 2.780 Cyclal C 180 2.301 Dimethyl Benzyl Carbinol 215 1.891 Dimethyl Benzyl Carbinol 215 1.891 Dimethyl Benzyl Carbinol 215 2.797 Ethyl Acetate 77 0.730 20 Ethyl Acetate 181 0.333 Ethyl Amyl Ketone 167 2.307 Ethyl Benzoate 212 2.640 Ethyl Butyrate 121 1.729 Ethyl Hexyl Ketone 190 2.916 2.916 Ethyl Phenyl Acetate 131 2.100 Ethyl Phenyl Acetate 229 2.489 Eucalyptol 176 2.756 Fenchyl Alcohol 200 2.579 30 Flor Acetate (tricyclo Decenyl Acetate 175 2.357 Frutene (tricyclo Decenyl Propionate) 200 2.260 Geraniol 40 Hexenol 159 1.397 Hexenyl Acetate 172 2.788 Hexyl Acetate 172 2.788 Hexyl Acetate 172 2.788 Hexyl Acetate 172 2.788 Hexyl Acetate 175 2.381 Hydratropic Alcohol 132 1.222 40 Isomenthone 210 2.831 Isoquinoline 243 2.080 Ligustral 177 2.301 Linalool 243 2.080 445 Linalool 207 2.650 Methyl Acetophenone 228 2.080 Methyl Amyl Ketone 152 2.848 425 4264 42	10	laevo-Carvone	230	2.203
cis-Jasmone 248 2.712 cis-3-Hexenyl Acetate 169 2.243 Cuminic alcohol 248 2.531 15 Cuminic aldehyde 236 2.780 Cyclal C 180 2.301 Dimethyl Benzyl Carbinol 215 1.891 Dimethyl Benzyl Carbinyl Acetate 250 2.797 Ethyl Acetate 77 0.730 20 Ethyl Aceto Acetate 181 0.333 Ethyl Amyl Ketone 167 2.307 Ethyl Benzoate 212 2.640 Ethyl Benzoate 212 2.640 Ethyl Benzoate 212 2.640 Ethyl Butyrate 121 1.729 Ethyl Hexyl Ketone 190 2.916 25 Ethyl-2-methyl butyrate 131 2.100 Ethyl Phenyl Acetate 122 2.489 Eucalyptol 176 2.756 Fenchyl Alcohol 200 2.579 30 Flor Acetate (tricyclo Decenyl Acetate) 175 2.357 <td></td> <td>Cinnamyl Formate</td> <td>250</td> <td>1.908</td>		Cinnamyl Formate	250	1.908
Cuminic alcohol 248 2.531		· ·	248	2.712
Cuminic alcohol 248 2.531		cis-3-Hexenyl Acetate	169	2.243
Cyclal C			248	2.531
Cyclal C Dimethyl Benzyl Carbinol 215 1.891 Dimethyl Benzyl Carbinyl Acetate 250 2.797 Ethyl Acetate 77 0.730 Ethyl Acetate 181 0.333 Ethyl Amyl Ketone 167 2.307 Ethyl Benzoate 212 2.640 Ethyl Butyrate 121 1.729 Ethyl Hexyl Ketone 190 2.916 Ethyl Methyl Pentanoate 143 2.700 Ethyl Methyl Pentanoate 143 2.700 Ethyl Phenyl Acetate 229 2.489 Eucalyptol 176 2.756 Fenchyl Alcohol 200 2.579 30 Flor Acetate (tricyclo Decenyl Acetate) 175 2.357 Frutene (tricyclo Decenyl Propionate) 200 2.260 Geraniol 230 2.649 Hexenol 159 1.397 Hexenyl Acetate 168 2.343 168 2.343 180 Hydroxycitronellal 241 1.541 1soamyl Alcohol 132 1.222 40 Isomenthone 210 2.831 Isoquinoline 243 2.080 Ligustral 177 2.301 Linalool 201 2.929 Methyl Acetophenone 228 2.080 Methyl Amyl Ketone 152 1.848 Methyl Amyl Ketone 152	15	Cuminic aldehyde	236	2.78 0
Dimethyl Benzyl Carbinyl Acetate 250 2.797 Ethyl Acetate 77 0.730 Ethyl Aceto Acetate 181 0.333 Ethyl Amyl Ketone 167 2.307 Ethyl Benzoate 212 2.640 Ethyl Butyrate 121 1.729 Ethyl Hexyl Ketone 190 2.916 Ethyl Hexyl Ketone 190 2.916 Ethyl Heyl Ketone 131 2.100 Ethyl Methyl Pentanoate 143 2.700 Ethyl Phenyl Acetate 229 2.489 Eucalyptol 176 2.756 Fenchyl Alcohol 200 2.579 For Acetate (tricyclo Decenyl Acetate) 175 2.357 Frutene (tricyclo Decenyl Propionate) 200 2.260 Geraniol 230 2.649 Hexenol 159 1.397 Hexenyl Acetate 168 2.343 35 Hexyl Acetate 168 2.343 35 Hexyl Acetate 172 2.787 Hexyl Formate 155 2.381 Hydratropic Alcohol 219 1.582 Hydroxycitronellal 241 1.541 Isoamyl Alcohol 132 1.222 40 Isomenthone 210 2.831 Isoquilegyl Acetate 239 2.100 Isoquinoline 243 2.080 Ligustral 177 2.301 Linalool 198 2.429 45 Linalool Oxide 188 1.575 Linalool Oxide 188 1.575 Linalool Oxide 188 1.575 Linalool Oxide 188 1.575 Linalyl Formate 202 2.929 Menthone 207 2.650 Methyl Amyl Ketone 152 1.848			180	2.301
Dimethyl Benzyl Carbinyl Acetate 250 2.797 Ethyl Acetate 77 0.730 Ethyl Aceto Acetate 181 0.333 Ethyl Amyl Ketone 167 2.307 Ethyl Benzoate 212 2.640 Ethyl Butyrate 121 1.729 Ethyl Hexyl Ketone 190 2.916 Ethyl-2-methyl butyrate 131 2.100 Ethyl Methyl Pentanoate 143 2.700 Ethyl Phenyl Acetate 229 2.489 Eucalyptol 176 2.756 Fenchyl Alcohol 200 2.579 Frutene (tricyclo Decenyl Acetate) 175 2.357 Frutene (tricyclo Decenyl Propionate) 200 2.649 Hexenol 159 1.397 Hexenyl Acetate 168 2.343 35 Hexyl Acetate 168 2.343 35 Hexyl Acetate 168 2.343 35 Hexyl Formate 155 2.381 Hydratropic Alcohol 219 1.582 Hydroxycitronellal 241 1.541 Isoamyl Alcohol 132 1.222 40 Isomenthone 210 2.831 Isopulegyl Acetate 239 2.100 Isoquinoline 243 2.080 Ligustral 177 2.301 Linalool 198 2.429 45 Linalool Oxide 188 1.575 Linalool Oxide 188 1.575 Linalool Oxide 188 1.575 Linalyl Formate 202 2.929 Menthone 207 2.650 Methyl Amyl Ketone 152 1.848 Methyl Amyl Ketone 152 1.848 Service		Dimethyl Benzyl Carbinol	215	1.891
Ethyl Acetate 777 0.730 Ethyl Aceto Acetate 181 0.333 Ethyl Amyl Ketone 167 2.307 Ethyl Benzoate 212 2.640 Ethyl Butyrate 121 1.729 Ethyl Hexyl Ketone 190 2.916 Ethyl-2-methyl butyrate 131 2.100 Ethyl Methyl Pentanoate 143 2.700 Ethyl Phenyl Acetate 229 2.489 Eucalyptol 176 2.756 Fenchyl Alcohol 200 2.579 30 Flor Acetate (tricyclo Decenyl Acetate) 175 2.357 Frutene (tricyclo Decenyl Propionate) 200 2.260 Geraniol 230 2.649 Hexenol 159 1.397 Hexenyl Acetate 168 2.343 35 Hexyl Acetate 168 2.343 36 Hexyl Acetate 172 2.787 Hexyl Formate 155 2.381 Hydratropic Alcohol 219 1.582 Hydroxycitronellal 241 1.541 Isoamyl Alcohol 132 1.222 40 Isomenthone 210 2.831 Isopulegyl Acetate 239 2.100 Isoquinoline 243 2.080 Ligustral 177 2.301 Linalool 198 2.429 45 Linalool Oxide 188 1.575 Linalyl Formate 202 2.929 Menthone 207 2.650 Methyl Acetophenone 228 2.080 Methyl Amyl Ketone 152 1.848			250	2.797
Ethyl Amyl Ketone Ethyl Benzoate Ethyl Benzoate Ethyl Butyrate Ethyl Hexyl Ketone Ethyl Hexyl Ketone 212 2.640 Ethyl Hexyl Ketone 190 2.916 25 Ethyl-2-methyl butyrate Ethyl Methyl Pentanoate Ethyl Phenyl Acetate Ethyl Phenyl Acetate Eucalyptol Fenchyl Alcohol 200 2.579 30 Flor Acetate (tricyclo Decenyl Acetate) Frutene (tricyclo Decenyl Propionate) Geraniol Hexenol Hexenol Hexenol Hexenyl Acetate 172 2.787 Hexyl Acetate Hexyl Formate Hydratropic Alcohol 150 Hydroxycitronellal Isoamyl Alcohol 132 1.541 Isoamyl Alcohol 132 1.222 40 Isomenthone 150 Isomenthone 210 Isomenthone 210 Isomenthone 210 Isomenthone 210 Isoquinoline 243 2.080 Ligustral Linalool 158 Linalool Oxide Ligustral Linalool 45 Linalool Oxide Ligustral Linalool Methyl Acetophenone Methyl Amyl Ketone 152 1.848 Methyl Amyl Ketone			77	0.730
Ethyl Amyl Ketone Ethyl Benzoate Ethyl Butyrate Ethyl Butyrate Ethyl Hexyl Ketone 212 Ethyl Hexyl Ketone 25 Ethyl-2-methyl butyrate Ethyl Methyl Pentanoate Ethyl Phenyl Acetate Ethyl Phenyl Acetate Ethyl Phenyl Acetate Ethyl Alcohol 30 Flor Acetate (tricyclo Decenyl Acetate) Frutene (tricyclo Decenyl Propionate) Geraniol Hexenol Hexenol Hexenyl Acetate Hexyl Formate Hexyl Formate Hydratropic Alcohol 155 Hydroxycitronellal Isoamyl Alcohol 132 1238 140 150 150 150 150 150 150 150 150 150 15	2 0	•	181	0.333
Ethyl Benzoate Ethyl Butyrate Ethyl Hexyl Ketone Ethyl Hexyl Ketone 25 Ethyl-2-methyl butyrate Ethyl Methyl Pentanoate Ethyl Phenyl Acetate (tricyclo Decenyl Acetate) Frutene (tricyclo Decenyl Acetate) Ethyl Propionate Ethyl Retard Ethyl Retard Ethyl Retard Ethyl Phenyl Acetate Ethyl Propionate Ethyl Retard Ethyl Retard Ethyl Retard Ethyl Retard Ethyl Retard Ethyl Ethy		•	167	2.307
Ethyl Butyrate			212	2.640
Ethyl Hexyl Ketone 190 2.916 25 Ethyl-2-methyl butyrate 131 2.100 Ethyl Methyl Pentanoate 143 2.700 Ethyl Phenyl Acetate 229 2.489 Eucalyptol 176 2.756 Fenchyl Alcohol 200 2.579 30 Flor Acetate (tricyclo Decenyl Acetate) 175 2.357 Frutene (tricyclo Decenyl Propionate) 200 2.260 Geraniol 230 2.649 Hexenol 159 1.397 Hexenyl Acetate 168 2.343 35 Hexyl Acetate 172 2.787 Hexyl Formate 155 2.381 Hydratropic Alcohol 219 1.582 Hydroxycitronellal 241 1.541 Isoamyl Alcohol 132 1.222 40 Isomenthone 210 2.831 Isoquinoline 243 2.080 Ligustral 177 2.301 Linalool Oxide 188 1.575 Linalyl Formate 202 2.929 Menthone 207 2.650 Methyl Acetophenone 228 2.080 Methyl Amyl Ketone 152 2.83		•	121	1.729
25 Ethyl-2-methyl butyrate 131 2.100 Ethyl Methyl Pentanoate 143 2.700 Ethyl Phenyl Acetate 229 2.489 Eucalyptol 176 2.756 Fenchyl Alcohol 200 2.579 30 Flor Acetate (tricyclo Decenyl Acetate) 175 2.357 Frutene (tricyclo Decenyl Propionate) 200 2.260 Geraniol 230 2.649 Hexenol 159 1.397 Hexenyl Acetate 168 2.343 35 Hexyl Acetate 172 2.787 Hexyl Formate 155 2.381 Hydraxycitronellal 219 1.582 Hydroxycitronellal 241 1.541 Isoamyl Alcohol 132 1.222 40 Isomenthone 210 2.831 Isoquinoline 243 2.080 Ligustral 177 2.301 Linalool 198 2.429 45 Linalool Oxide 188 1.575 </td <td></td> <td></td> <td>190</td> <td>2.916</td>			190	2.916
Ethyl Methyl Pentanoate 143 2.700 Ethyl Phenyl Acetate 229 2.489 Eucalyptol 176 2.756 Fenchyl Alcohol 200 2.579 30 Flor Acetate (tricyclo Decenyl Acetate) 175 2.357 Frutene (tricyclo Decenyl Propionate) 200 2.260 Geraniol 230 2.649 Hexenol 159 1.397 Hexenyl Acetate 168 2.343 35 Hexyl Acetate 172 2.787 Hexyl Formate 155 2.381 Hydratropic Alcohol 219 1.582 Hydroxycitronellal 241 1.541 Isoamyl Alcohol 132 1.222 40 Isomenthone 210 2.831 Isoquinoline 243 2.080 Ligustral 177 2.301 Linalool 198 2.429 45 Linalool Oxide 188 1.575 Linalyl Formate 202 2.929 Menthone 207 2.650 Methyl Acetophenone 228<	25	•	131	2.100
Eucalyptol			143	2.700
Fenchyl Alcohol 200 2.579 Fenchyl Alcohol 200 2.357 Frutene (tricyclo Decenyl Acetate) 175 2.357 Frutene (tricyclo Decenyl Propionate) 200 2.260 Geraniol 230 2.649 Hexenol 159 1.397 Hexenyl Acetate 168 2.343 35 Hexyl Acetate 172 2.787 Hexyl Formate 155 2.381 Hydratropic Alcohol 219 1.582 Hydroxycitronellal 241 1.541 Isoamyl Alcohol 132 1.222 40 Isomenthone 210 2.831 Isopulegyl Acetate 239 2.100 Isoquinoline 243 2.080 Ligustral 177 2.301 Linalool 198 2.429 45 Linalool Oxide 188 1.575 Linalyl Formate 202 2.929 Menthone 207 2.650 Methyl Acetophenone 228 2.080 Methyl Amyl Ketone 152 1.848		Ethyl Phenyl Acetate	229	
Flor Acetate (tricyclo Decenyl Acetate) 175 2.357 Frutene (tricyclo Decenyl Propionate) 200 2.260 Geraniol 230 2.649 Hexenol 159 1.397 Hexenyl Acetate 168 2.343 2.787 Hexyl Formate 155 2.381 Hydratropic Alcohol 219 1.582 Hydroxycitronellal 241 1.541 Isoamyl Alcohol 132 1.222 40 Isomenthone 210 2.831 Isopulegyl Acetate 239 2.100 Isoquinoline 243 2.080 Ligustral 177 2.301 Linalool 198 2.429 45 Linalool Oxide 188 1.575 Linalyl Formate 202 2.929 Menthone 207 2.650 Methyl Acetophenone 228 2.080 Methyl Amyl Ketone 152 1.848		Eucalyptol	176	2.756
Frutene (tricyclo Decenyl Propionate) 200 2.260 Geraniol 230 2.649 Hexenol 159 1.397 Hexenyl Acetate 168 2.343 35 Hexyl Acetate 172 2.787 Hexyl Formate 155 2.381 Hydratropic Alcohol 219 1.582 Hydroxycitronellal 241 1.541 Isoamyl Alcohol 132 1.222 40 Isomenthone 210 2.831 Isopulegyl Acetate 239 2.100 Isoquinoline 243 2.080 Ligustral 177 2.301 Linalool 198 2.429 45 Linalool Oxide 188 1.575 Linalyl Formate 202 2.929 Menthone 207 2.650 Methyl Acetophenone 228 2.080 Methyl Amyl Ketone 152 1.848		Fenchyl Alcohol		
Geraniol 159 1.397 Hexenol 168 2.343 Hexenyl Acetate 168 2.343 Hexyl Acetate 172 2.787 Hexyl Formate 155 2.381 Hydratropic Alcohol 219 1.582 Hydroxycitronellal 241 1.541 Isoamyl Alcohol 132 1.222 Isomenthone 210 2.831 Isopulegyl Acetate 239 2.100 Isoquinoline 243 2.080 Ligustral 177 2.301 Linalool 198 2.429 Linalool Oxide 188 1.575 Linalyl Formate 202 2.929 Menthone 207 2.650 Methyl Acetophenone 228 2.080 Methyl Amyl Ketone 152 1.848	30	Flor Acetate (tricyclo Decenyl Acetate)	175	
Hexenol 159 1.397 Hexenyl Acetate 168 2.343 35 Hexyl Acetate 172 2.787 Hexyl Formate 155 2.381 Hydratropic Alcohol 219 1.582 Hydroxycitronellal 241 1.541 Isoamyl Alcohol 132 1.222 40 Isomenthone 210 2.831 Isopulegyl Acetate 239 2.100 Isoquinoline 243 2.080 Ligustral 177 2.301 Linalool 198 2.429 45 Linalool Oxide 188 1.575 Linalyl Formate 202 2.929 Menthone 207 2.650 Methyl Acetophenone 228 2.080 Methyl Amyl Ketone 152 1.848		Frutene (tricyclo Decenyl Propionate)		
Hexenyl Acetate 168 2.343 35 Hexyl Acetate 172 2.787 Hexyl Formate 155 2.381 Hydratropic Alcohol 219 1.582 Hydroxycitronellal 241 1.541 Isoamyl Alcohol 132 1.222 40 Isomenthone 210 2.831 Isopulegyl Acetate 239 2.100 Isoquinoline 243 2.080 Ligustral 177 2.301 Linalool 198 2.429 45 Linalool Oxide 188 1.575 Linalyl Formate 202 2.929 Menthone 207 2.650 Methyl Acetophenone 228 2.080 Methyl Amyl Ketone 152 1.848		Geraniol		
Hexyl Acetate		Hexenol		
Hexyl Formate		Hexenyl Acetate		
Hydratropic Alcohol 219 1.582 Hydroxycitronellal 241 1.541 Isoamyl Alcohol 132 1.222 40 Isomenthone 210 2.831 Isopulegyl Acetate 239 2.100 Isoquinoline 243 2.080 Ligustral 177 2.301 Linalool 198 2.429 45 Linalool Oxide 188 1.575 Linalyl Formate 202 2.929 Menthone 207 2.650 Methyl Acetophenone 228 2.080 Methyl Amyl Ketone 152 1.848	35	Hexyl Acetate		
Hydroxycitronellal 241 1.541 Isoamyl Alcohol 132 1.222 40 Isomenthone 210 2.831 Isopulegyl Acetate 239 2.100 Isoquinoline 243 2.080 Ligustral 177 2.301 Linalool 198 2.429 45 Linalool Oxide 188 1.575 Linalyl Formate 202 2.929 Menthone 207 2.650 Methyl Acetophenone 228 2.080 Methyl Amyl Ketone 152 1.848		Hexyl Formate		
Isoamyl Alcohol 132 1.222 1.222 1.222 1.222 1.222 1.222 1.222 1.222 1.222 1.222 1.222 1.222 1.222 1.200 1.20		Hydratropic Alcohol		
40 Isomenthone 210 2.831 Isopulegyl Acetate 239 2.100 Isoquinoline 243 2.080 Ligustral 177 2.301 Linalool 198 2.429 2.429 45 Linalool Oxide 188 1.575 Linalyl Formate 202 2.929 Menthone 207 2.650 Methyl Acetophenone 228 2.080 Methyl Amyl Ketone 152 1.848 2.848 2.884 2.88		Hydroxycitronellal		
Isopulegyl Acetate 239 2.100 Isoquinoline 243 2.080 Ligustral 177 2.301 Linalool 198 2.429 45 Linalool Oxide 188 1.575 Linalyl Formate 202 2.929 Menthone 207 2.650 Methyl Acetophenone 228 2.080 Methyl Amyl Ketone 152 1.848		Isoamyl Alcohol		
Isoquinoline 243 2.080 Ligustral 177 2.301 Linalool 198 2.429 2.429 2.429 2.429 2.429 2.429 2.929 2.929 2.929 2.650 Methyl Acetophenone 228 2.080 Methyl Amyl Ketone 252 2.828 2.080	40	Isomenthone		
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Linalool 198 2.429 45 Linalool Oxide 188 1.575 Linalyl Formate 202 2.929 Menthone 207 2.650 Methyl Acetophenone 228 2.080 Methyl Amyl Ketone 152 1.848				
45 Linalool Oxide 188 1.575 Linalyl Formate 202 2.929 Menthone 207 2.650 Methyl Acetophenone 228 2.080 Methyl Amyl Ketone 152 1.848				
Linalyl Formate 202 2.929 Menthone 207 2.650 Methyl Acetophenone 228 2.080 Methyl Amyl Ketone 152 1.848				
Menthone 207 2.650 Methyl Acetophenone 228 2.080 Methyl Amyl Ketone 152 1.848	45			
Methyl Acetophenone 228 2.080 Methyl Amyl Ketone 152 1.848		•		
Methyl Amyl Ketone 152 1.848				
207		-		
50 Methyl Anthranilate 237 2.024				
	5 0	Methyl Anthranilate	237	2.024

	Methyl Benzoate	200	2.111
	Methyl Benzyl Acetate	213	2.300
	Methyl Eugenol	249	2.783
	Methyl Heptenone	174	1.703
5	Methyl Heptine Carbonate	217	2.528
	Methyl Heptyl Ketone	194	1.823
	Methyl Hexyl Ketone	173	2.377
	Methyl Phenyl Carbinyl Acetate	214	2.269
	Methyl Salicylate	223	1.960
10	Nerol	227	2.649
	Octalactone	230	2.203
	Octyl Alcohol (Octanol-2)	179	2.719
	para-Cresol	202	1.000
	para-Cresyl Methyl Ether	176	2.560
15	para-Methyl Acetophenone	228	2.080
	Phenoxy Ethanol	245	1.188
	Phenyl Acetaldehyde	195	1.780
	Phenyl Ethyl Acetate	232	2.129
	Phenyl Ethyl Alcohol	220	1.183
20	Phenyl Ethyl Dimethyl Carbinol	238	2.420
	Prenyl Acetate	155	1.684
	Propyl Butyrate	143	2.210
	Pulegone	224	2.350
	Rose Oxide	182	2.896
25	Safrole	234	1.870
	4-Terpinenol	212	2.749
	alpha-Terpineol	219	2.569
	Viridine	221	1.293

TABLE 2. Perfume materials with boiling point less than or equal to 250 °C and CLogP greater than or equal to 3.0

	Name	BP	CLogP
35	Zallo-Ocimene	192	4.362
	Allyl Heptoate	210	3.301
	Anethol	236	3.314
	Benzyl Butyrate	240	3.698
	Camphene	159	4.192
40	Carvacrol	238	3.401
	cis-3-Hexenyl Tiglate	101	3.700
	Citral (Neral)	228	3.120
	Citronellol	225	3.193
	Citronellyl Acetate	229	3.670
45	Citronellyl Isobutyrate	249	4.937
	Citronellyl Nitrile	225	3.094
	Citronellyl Propionate	242	4.628
	Cyclohexyl Ethyl Acetate	187	3.321
	Decyl Aldehyde	209	4.008
50	Delta Damascone	242	3.600

	Dihydro Myrcenol	208	3.030
	Dihydromyrcenyl Acetate	225	3.879
	Dimethyl Octanol	213	3.737
	Fenchyl Acetate	220	3.485
5	gamma Methyl Ionone	230	4.089
	gamma-Nonalactone	243	3.140
	Geranyl Acetate	245	3.715
	Geranyl Formate	216	3.269
	Geranyl Isobutyrate	245	4.393
10	Oeranyl Nitrile	222	3.139
	Hexenyl Isobutyrate	182	3.181
	Hexyl Neopentanoate	224	4.374
	Hexyl Tiglate	231	3.800
	alpha-Ionone	237	3.381
15	beta-Ionone	239	3.960
	gamma-Ionone	240	3.780
	alpha-Irone	250	3.820
	Isobornyl Acetate	227	3.485
	Isobutyl Benzoate	242	3.028
20	Isononyl Acetate	200	3.984
20	Isononyl Alcohol	194	3.078
	Isomenthol	219	3.030
	para-Isopropyl Phenylacetaldehyde	243	3.211
	Isopulegol	212	3.330
25	Lauric Aldehyde (Dodecanal)	249	5.066
23	d-Limonene	177	4.232
	Linalyl Acetate	220	3.500
	Menthyl Acetate	227	3.210
	Methyl Chavicol	216	3.074
30	alpha-iso "gamma" Methyl Ionone	230	4.209
30	Methyl Nonyl Acetaldehyde	232	4.846
	Methyl Octyl Acetaldehyde	228	4.317
	Myrcene	167	4.272
	Neral	228	3.320
25		231	3.555
35	Neryl Acetate	212	4.374
	Nonyl Acetate Nonyl Aldehyde	212	3.479
	•	223	3.845
	Octyl Aldehyde		4.232
40	Orange Terpenes (d-Limonene)	177	4.232
40	para-Cymene	. 179	3.000
	Phenyl Ethyl Isobutyrate	250	4.122
	alpha-Pinene	157	4.122
	beta-Pinene	166	4.162
	alpha-Terpinene	176	4.232
45	gamma-Terpinene	183	
	Terpinolene	184	4.232
	Terpinyl acetate	220	3.475
	Tetrahydro Linalool	191	3.517
	Tetrahydro Myrcenol	208	3.517
50	Undecenal	223	4.053

Veratrol	206	3.140
Verdox	221	4.059
Vertenex	232	4.060

<u>TABLE 3</u>. Perfume materials with boiling point greater than 250 $^{\circ}$ C

	Name	BP	CLogP
	Allyl Cyclohexane Propionate	267	3.935
10	Ambrettolide	300	6.261
	Amyl Benzoate	262	3.417
	Amyl Cinnamate	310	3.771
	Amyl Cinnamic Aldehyde	285	4.324
	Amyl Cinnamic Aldehyde Dimethyl Acetal	300	4.033
15	iso-Amyl Salicylate	277	4.601
	Aurantiol	450	4.216
	Benzophenone	306	3.120
	Benzyl Salicylate	300	4.383
	Cadinene	275	7.346
20	Cedro1	291	4.530
	Cedryl Acetate	303	5.436
	Cinnamyl Cinnamate	370	5.480
	Coumarin	291	1.412
	Cyclohexyl Salicylate	304	5.265
25	Cyclamen Aldehyde	270	3.680
	Dihydro Isojasmonate	300	3.009
	Diphenyl Methane	262	4.059
	Ethylene Brassylate	332	4.554
	Ethyl Methyl Phenyl Glycidate	260	3.165
30	Ethyl Undecylenate	264	4.888
	iso-Eugenol	266	2.547
	Exaltolide	280	5.346
	Galaxolide	260	5.482
	Oeranyl Anthranilate	312	4.216
35	Hexadecanolide	294	6.805
	Hexenyl Salicylate	271	4.716
	Hexyl Cinnamic Aldehyde	305	5.473
	Hexyl Salicylate	290	5.260
	Linalyl Benzoate	263	5.233
40	2-Methoxy Naphthalene	275	3.235
	Methyl Cinnamate	263	2.620
	Methyl Dihydrojasmonate	300	2.275
	beta-Methyl Naphthyl ketone	300	2.275
	Musk Indanone	>250	5.458
45	Musk Ketone	137 (MP)	3.014
	Musk Tibetine	136 (MP)	3.831
	Myristicin	276	3.200
	delta-Nonalaetone	280	2.760
	Oxahexadeeanolide-10	300	4.336
50	Oxahexadecanolide-ll	35 (MP)	4.336

	Patchouli Aleohol	285	4.530
	Phantolide	288	5.977
	Phenyl Ethyl Benzoate	300	4.058
	Phenylethylphenylacetate	325	3.767
5	alpha-Santalol	301	3.800
	Thibetolide	280	6.246
	delta-Undecalaetone	290	3.830
	gamma-Undecalacto ne	297	4.140
	Vanillin	285	1.580
10	Vetiveryl Acetate	285	4.882
	Yara-Yara	274	3.235

Hair Care Preparations

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The silicone-in-water emulsion, wherein the dispersed silicone has a viscosity of at least, and preferably more than 100 million mm²s⁻¹, and a particle size of at most, preferably less than 2 μm, is mixed with a fragrance composition in a weight ratio of preferably from 1:1 to 10:1 of silicone:fragrance. It is preferred that this mixing takes place prior to incorporation in a hair care preparation, as it enables a greater control and consistency of the distribution and partitioning of the silicone and the fragrance in the hair care preparation. Mixing may take place by methods well-known in the art. For example, low shear mixing may take place by methods well-known in the art, as for example, regular hand mixing using low shear bench scale overhead mixing equipment, as for example a Heidolf RZR 50 stirrer with variable speed range between 45-2000 RPM, or on a larger scale in a groin kettle or via large scale overhead mechanical drum mixing.

The mixture is then incorporated in a hair care preparation in an amount which provides at least 1% silicone by weight of the hair care preparation, for example 1 to 10% by weight, or for example 2 to 6% wt. Hair care preparations, including shampoos, conditioners and conditioning shampoos are well known in the art, and typically comprise a washing base containing surfactants, and other agents for

improving the cosmetic properties of the preparation. The washing base typically comprises a cosmetically acceptable aqueous medium, which may be solely water, or water with a solvent (for example, a lower alcohol or alkylene glycol). The surfactants may be anionic, non-ionic, cationic, or amphoteric. Other additives which may be included in the preparation include pH adjusters (for example, amines), viscosity regulators, electrolytes or thickeners, pearlising or opacifying agents, foam synergists, and other agents for improving the properties of the preparation, for example anionic, non-ionic and cationic polymers, proteins, fatty acids and vitamins. The hair care preparation will typically be in the form of a liquid, a thickened liquid, a lotion or a gel.

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The invention is directed to a composition, wherein the composition is a hair care preparation, comprising: i) a fragrance composition; and ii) a silicone-inwater emulsion comprising at least one surfactant, wherein the dispersed silicone has a viscosity of 100 million mm²s⁻¹ or more and a particle size of less than 2 μm. The invention is further directed to a composition, wherein the composition is a hair care preparation, comprising: i) a fragrance composition, wherein the fragrance composition comprises a) at least 10.0% by weight of a fragrance composition of one or more perfume ingredients having a boiling point of 250 °C or less and ClogP of less than 3.0; b) at least 15% by weight of a fragrance composition of one or more perfume ingredients having a boiling point of 250 °C or less and ClogP of 3.0 or more; and c) at least 20% by weight of a fragrance composition of one or more perfume ingredients have a boiling point of greater than 250 °C; and ii) a silicone-in-water emulsion comprising at least one surfactant. The invention is further directed to a composition. wherein the composition is a hair care preparation, comprising: i) a fragrance composition, wherein the fragrance composition comprises a) at least 10.0% by weight of a fragrance composition of one or more perfume ingredients having a boiling

point of 250 °C or less and ClogP of less than 3.0; b) at least 20% by weight of a fragrance composition of one or more perfume ingredients having a boiling point of 250 °C or less and ClogP of 3.0 or more; and c) at least 30% by weight of a fragrance composition of one or more perfume ingredients have a boiling point of greater than 250 °C; and ii) a silicone-in-water emulsion comprising at least one surfactant. In one embodiment, the silicone:fragrance composition weight ratio is from 1:1 to 10:1. In a further embodiment, the hair care preparation is a shampoo, a conditioner, or a conditioning shampoo.

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The use of the above defined silicone-in-water emulsion in hair care

preparations according to the present invention can provide hair care preparations

which have improved fragrance deposition on hair, measured as longer lasting

fragrance release from the hair compared to conventional preparations. The improved

fragrance deposition occurs without a loss of favorable sensory and aesthetic perception

for the user normally associated with silicone deposition on hair, which includes ease of

combing, luster, shine and general conditioning.

It is particularly surprising that the combination of such high viscosity of the emulsified silicones with such low particle size results in the greater deposition of the fragrance or a longer lasting release of the fragrance from the hair, without negatively affecting the benefits of deposited silicones.

The invention is directed to a method of cleansing hair, which comprises applying to the hair a hair care preparation comprising: i) a fragrance composition; and ii) a silicone-in-water emulsion comprising at least one surfactant, wherein the dispersed silicone has a viscosity of 100 million mm²s⁻¹ or more and a particle size of less than 2 µm.

The invention is also directed to a method of cleansing hair, which comprises applying to the hair a hair care preparation comprising: i) a fragrance composition, wherein the fragrance composition comprises a) at least 10.0% by weight of a fragrance composition of one or more perfume ingredients having a boiling point of 250 °C or less and ClogP of less than 3.0; b) at least 15% by weight of a fragrance composition of one or more perfume ingredients having a boiling point of 250 °C or less and ClogP of 3.0 or more; and c) at least 20% by weight of a fragrance composition of one or more perfume ingredients have a boiling point of greater than 250 °C; and ii) a silicone-in-water emulsion comprising at least one surfactant.

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The invention is also directed to a method of cleansing hair, which comprises applying to the hair a hair care preparation comprising: i) a fragrance composition, wherein the fragrance composition comprises a) at least 10.0% by weight of a fragrance composition of one or more perfume ingredients having a boiling point of 250 °C or less and ClogP of less than 3.0; b) at least 20% by weight of a fragrance composition of one or more perfume ingredients having a boiling point of 250 °C or less and ClogP of 3.0 or more; and c) at least 30% by weight of a fragrance composition of one or more perfume ingredients have a boiling point of greater than 250 °C; and ii) a silicone-in-water emulsion comprising at least one surfactant.

In a preferred embodiment, the silicone: fragrance composition weight ratio is from 1:1 to 10:1. In a further preferred embodiment, the hair care preparation is a shampoo, a conditioner, or a conditioning shampoo.

The compositions of the invention may additionally be used in other cosmetic applications, in particular skin care applications, providing a similar benefit. Examples of suitable skin care applications include lotions, gels (including shower gels), jellies, pastes, facial applications, moisturizers, face masks, night creams, day

creams, sun milk, massage creams, make-up removal lotions, anti-aging cream, and anti-wrinkle cream.

The following Examples are provided for a further illustration of the invention, but are not intended to be limiting thereof. Unless otherwise indicated, all parts and percentages are by weight and all viscosities are dynamic viscosities measured at 25 °C.

Examples

10 Example 1

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A silicone-in-water emulsion, SEM1, was made by mixing 58.5 parts dimethylvinyl-terminated polydimethylsiloxane having a viscosity of 55,000 mm²s⁻¹ with 1.17 parts of dimethylhydrogen-terminated polydimethylsiloxane having a viscosity of 10 mm²s⁻¹. To this mixture was added 2.05 parts of polyoxyethylene (3) lauryl ether and enough water to cause phase inversion. 2.8 parts of polyoxyethylene (23) lauryl ether were then added under high shear. Finally, water was added to a total of 35.5 parts water, in which the water was added in stages with mixing after each addition, followed by the addition of 0.012 parts of a platinum catalyst and 0.5 parts of Glydant Plus preservative. The resulting emulsion was a non-ionic emulsion containing about 60% by weight of a linear silicone polymer formed from the hydrosilylation of a Si-H end-blocked polydimethyl siloxane and a vinyl-end-blocked polydimethyl siloxane, in which the polymer has a viscosity of about 170 million mm²s⁻¹ and an average particle size of 0.5 µm. The silicone-in-water emulsion formed, SEM1, was mixed with fragrances and tested as described in the following Examples.

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Example 2

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Fragrance 862883 was mixed with the silicone-in-water emulsion, SEM1, in a 1:4 ratio by weight. The fragrance/silicone mixture was added at a concentration of 4% by weight to a shampoo base T-058.

Shampoo base T-058 is formed by mixing 0.30 % Metholcel E4M (hydroxypropyl methylcellulose; Dow Chemical Co.) in hot deionized water at 80 - 90 °C with an overhead mixer at medium speed until all of the particles are wetted down and a consistent dispersion is obtained. 0.05 % Sequestrene Na4 (tetrasodium EDTA; Ciba Geigy) was mixed at 85 °C with deionized water, was cooled down, and added to the hot mixture. The resulting solution was cooled to 30 °C and mixing was continued for approximately 20 minutes. The mixture was reheated to 60 °C. 18.00 % Standapol T (TEA lauryl sulfate; Cognis/Henkel), 8.00 % Standapol A (ammonium lauryl sulfate; Cognis/Henkel), 4.00 % Monamid 150-LMWC (lauramide DEA; Uniqema), and 0.30 % Palmitic acid (Ruger) were added while mixing at low/medium speed using an overhead mixer. 0.15 % Gydant (DMDM hydantoin; Lonza), approximately 0.3 % sodium chloride (to adjust viscosity), and approximately 0.5 % citric acid (to adjust pH to 5.5-6.0) was added while mixing at low/medium speed using an overhead mixer. The mixture was mixed at low speed until room temperature was reached to form T-058.

As a control, fragrance alone was added at a concentration of 0.8% by weight to the same shampoo base T-058. The fragrance/silicone mixture and the control each were added to half a head of two subjects. The hair was shampooed and the fragrance was evaluated by a panel of five perfume and odor evaluators at the stages of neat products, hair during lathering, hair after rinsing, hair after blow drying, and hair 6 hours after drying. The evaluations were determined on a scale of 1-5 (1, no

fragrance, to 5, strong fragrance smell) for each subject. The evaluations for the two subjects were averaged and the evaluation results are summarized in Table 4.

5 TABLE 4. Effect of silicone on fragrance performance of hair care preparation.

Stage	4% fragrance/silicone mixture in shampoo base T- 058	0.8 % fragrance mixture in shampoo base T-058
Neat	4.00	4.25
Lathering	3.25	2.75
Rinse	2.75	1.50
Blow-dry	2.50	1.75
6 hours after blow-dry	2.00	1.50

The results indicate that silicones have a significant effect on residual fragrance perception at the stages of lathering, rinse, blow-dry, and 6 hours after blow-dry.

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Example 3

Fragrance 862883 was mixed with the silicone-in-water emulsion, SEM1, in a 1:4 ratio by weight. The fragrance/silicone mixture was added at a concentration of 4% by weight to a shampoo base T-058 (see Example 2). As a control, fragrance alone was added at a concentration of 0.8% by weight to the same shampoo base T-058. The fragrance/silicone mixture and the control were added to 6 swatches of hair each. The 12 hair swatches were shampooed and the fragrance was

evaluated by a panel of five perfume and odor evaluators at the stages of hair after rinsing, hair after blow drying, and hair 6 hours after drying. The evaluations were determined on a scale of 1-5 (1, no fragrance, to 5, strong fragrance smell) for each swatch. The evaluations by each evaluator and for all the swatches were averaged, and the evaluation results are indicated in Table 5.

<u>TABLE 5</u>. Effect of silicone on fragrance performance of hair care preparation.

Stage	4% fragrance/silicone	0.8 % fragrance mixture in
	mixture in shampoo base T-	shampoo base T-058
	058	
Rinse	3.60	3.26
Blow-dry	3.02	2.75
6 hours after blow-dry	1.93	2.13

The results indicate that silicones have a positive effect on residual

fragrance perception at the stages of after rinse and after blow-dry.

Example 4

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Fragrance 890656 was mixed with the silicone-in-water emulsion, SEM1, in a 1:8 ratio by weight. The fragrance/silicone mixture was added at a concentration of 3.2% by weight to a conditioner base T-114.

Conditioner base T-114 is formed by heating deionized water to 70 °C and slowing adding 0.25 % methyl paraben (TRI-K) while mixing at medium speed until completely solubilized. The resulting composition is held at 70 °C. 1.50 % propylene glycol and 1.00 % Natrosol 250 HR (hydroxythylcellulose; Hercules) are

at 70 °C until completely clear and hydrated. 1.00 % Ammonyx 4002 (stearalkonium chloride; Stephan) is slowly added to the mixture. 0.25 % Lipoquat R (ricinoleamidopropyl ethyldimonium ethosulfate; Lipo Chemicals, Inc.), 0.50 % Lipopeg 6000 DS (PEG-150 distearate; Lipo Chemicals, Inc.), 1.00 % Lipowax D (cetearyl alcohol and ceteareth-20; Lipo Chemicals, Inc.), and 0.10 % propyl paraben

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(cetearyl alcohol and ceteareth-20; Lipo Chemicals, Inc.), and 0.10 % propyl paraben (TRI-K) were pre-heated until melted and clear and then added to the mixture while mixing on a homomixer at medium speed for 1 minute to form conditioner base T-114.

As a control, fragrance alone was added at a concentration of 0.4% by weight to the same conditioner base T-114. The fragrance/silicone mixture and the control were added to 6 swatches of hair each. The 12 hair swatches were conditioned and the fragrance was evaluated by a panel of five perfume and odor evaluators at the stages of hair after rinsing, hair after blow drying, and hair 6 hours after drying. The evaluations were determined on a scale of 1-5 (1, no fragrance, to 5, strong fragrance smell) for each swatch. The evaluations by each evaluator and for all the swatches were averaged and the evaluation results are indicated in Table 6.

<u>TABLE 6</u>. Effect of silicone on fragrance performance of hair care preparation.

Stage	3.2% fragrance/silicone mixture in conditioner base	0.4 %fragrance mixture in conditioner base T-114
	T-114	
Rinse	3.23	2.83
Blow-dry	2.68	2.42
6 hours after blow-dry	2.03	2.00

The results indicate that silicones have a positive effect on residual fragrance perception at the stages of after rinse and after blow-dry.

Example 5

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Fragrance 862883 was mixed with the silicone-in-water emulsion, SEM2. SEM2 was prepared in a similar manner as SEM1 described in Example 1, except that the non-ionic surfactants were replaced with 3.4 parts of Bioterge AS 40 (Stephan Company) and less dimethylhydrogen-terminated polydimethylsiloxane was used, resulting in an anionic emulsion having a polymer with a viscosity of 300 million mm²s⁻¹ and an average particle size of 1.1 µm. The emulsion and the fragrance were mixed in a 4:1 ratio by weight. The SEM2-based fragrance/silicone mixture was added at a concentration of 4% by weight to a shampoo base T-058 (see Example 2).

Fragrance 862883 was mixed with the silicone-in-water emulsion, SEM8. SEM8 was prepared in a similar manner as SEM1 described in Example 1, except that in addition, 4 parts of a alkylmethyl siloxane polymer of about 10 mm²s⁻¹ having a C₃₀ alkyl substituent on each silicon atom, was added. The emulsion had an average particle size of 0.8 μm. The emulsion was mixed with fragrance in a 4:1 ratio by weight. The SEM8-based fragrance/silicone mixture was added at a concentration of 4% by weight to a shampoo base T-058.

As a control, fragrance alone was added at a concentration of 0.8% by weight to the same shampoo base T-058. The fragrance/silicone mixtures and the control were added to 3 swatches of hair each. The 9 hair swatches were shampooed and the fragrance was evaluated by a panel of six perfume and odor evaluators at the stages of hair after rinsing, hair after blow drying, and hair 6 hours after drying. The evaluations were determined on a scale of 1-5 (1, no fragrance, to 5, strong fragrance

smell) for each swatch. The evaluations by each evaluator and for all the swatches were averaged and the evaluation results are indicated in Table 7.

TABLE 7. Effect of silicone on fragrance performance of hair care preparation.

Stage	4% SEM2-based fragrance/silicone mixture in shampoo base T-058	4% SEM8-based fragrance/silicone mixture in shampoo base T-058	0.4 % fragrance mixture in shampoo base T- 058
Rinse	2.82	3.44	3.29
Blow-dry	2.00	2.71	2.74
6 hours after blow-dry	1.58	1.94	1.98

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The results indicate that the SEM2 and SEM8-based mixtures do not have a consistent positive effect on residual fragrance perception.

Example 6

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Fragrance 862883 was mixed with the silicone-in-water emulsion, SEM1, in a 1:4 ratio by weight. The SEM1-based fragrance/silicone mixture was added at a concentration of 4% by weight to a shampoo base T-058 (see Example 2).

Fragrance 862883 was mixed with the silicone-in-water emulsion, SEM4, in a 1:4 ratio by weight. SEM4 was made in a similar way to SEM1 described in Example 1, however the resulting emulsion has a viscosity of 138 million $\mathrm{mm}^2\mathrm{s}^{\text{-}1}$ and an average particle size of $0.35~\mu m$. The SEM4-based fragrance/silicone mixture was added at a concentration of 4% by weight to a shampoo base T-058.

Fragrance 862883 was mixed with the silicone-in-water emulsion, SEM5, in a 1:4 ratio by weight. SEM5 was made in a similar fashion as SEM1 described in Example 1, but the resulting emulsion had a viscosity of 272 million mm 2 s $^{-1}$ and an average particle size of 0.36 μ m. The SEM5-based fragrance/silicone mixture was added at a concentration of 4% by weight to a shampoo base T-058.

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As a control, fragrance alone was added at a concentration of 0.8% by weight to the same shampoo base T-058. The fragrance/silicone mixtures and the control were added to 3 swatches of hair each. The 12 hair swatches were shampooed and the fragrance was evaluated by a panel of six perfume and odor evaluators at the stages of hair after rinsing, hair after blow drying, and hair 6 hours after drying. The evaluations were determined on a scale of 1-5 (1, no fragrance, to 5, strong fragrance smell) for each swatch. The evaluations by each evaluator and for all the swatches were averaged and the evaluation results are indicated in Table 8.

15 TABLE 8. Effect of silicone on fragrance performance of hair care preparation.

Stage	4% SEM1-based	4% SEM4-based	4% SEM5-based	0.8 %
	fragrance/silicone	fragrance/silicone	fragrance/silicone	fragrance
	mixture in	mixture in	mixture in	mixture in
	shampoo base T-	shampoo base T-	shampoo base T-	shampoo
	058	058	058	base T-058
Rinse	3.72	3.36	3.58	3.10
Blow-	3.43	2.97	3.07	2.92
dry				
6 hours	2.11	1.83	2.03	1.64
after				
blow-				
dry				

The results indicate that the SEM1, SEM4, and SEM5-based mixtures all give a positive effect on residual fragrance perception. Furthermore, SEM1-based mixture give a better effect than SEM4 or SEM5-based mixtures.

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Example 7

Compositions of shampoo base T-058 alone or in combination with silicone-in-water emulsion SEM1 and/or fragrance 862883 were formulated according to the ingredients and amounts given in Table 9. The viscosity of the resulting compositions are shown in Table 9. Addition of silicone-in-water emulsion SEM1 to shampoo base T-058 reduced the viscosity of the shampoo base T-058 by about 15%. Addition of fragrance 862883 increased the viscosity of shampoo base T-058 over 2.5-fold. When added to shampoo base T-058, the combination of silicone-in-water emulsion SEM1 and fragrance 862883 exhibited a viscosity reduction of about 15%, relative to addition of fragrance 862883 alone.

TABLE 9. Effect of silicone on viscosity of shampoo-fragrance compositions

Composition	Viscosity (cps)	
Shampoo base T-058	2020	
Shampoo base T-058 + 3.2% silicone-in- water emulsion SEM1	1700	
Shampoo base T-058+ 0.8% fragrance 862883	5450	
Shampoo base T-058 + 3.2% silicone-in- water emulsion SEM1 + 4.0% fragrance	4530	
862883		

The present invention is not to be limited in scope by the specific embodiments described herein. Various modifications of the invention in addition to those described herein will become apparent to those skilled in the art from the foregoing description. Such modifications are intended to fall within the scope of the appended claims.

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All patents, applications, publications, test methods, literature, and other materials cited herein are hereby incorporated by reference.

WE CLAIM:

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- 1. A composition comprising:
 - i) a fragrance composition; and
- ii) a silicone-in-water emulsion comprising at least one surfactant, wherein the dispersed silicone has a viscosity of 100 million mm²s⁻¹ or more and a partiele size of 2 μ m or less.
- 2. The composition of claim 1, wherein the silicone-in-water emulsion is non-ionic and the silicone is a linear polysiloxane polysilalkylene copolymer, wherein the monovalent silicone substituents are substantially all C_{1-6} hydrocarbon groups and the copolymer has a viscosity of 150 million mm²s⁻¹ or more and a particle size of 1 μ m or less.
- 3. The composition of claim 2, wherein the silicone-in-water emulsion is a non-ionic emulsion comprising about 60 % by weight of a linear silicone copolymer formed by the hydrosilylation of a Si-H end-blocked polydimethyl siloxane and a vinyl-end-blocked polydimethyl siloxane, wherein the copolymer has a viscosity of about 170 mm²s⁻¹ and a particle size of about 0.5 μ m, and about 5 % by weight of non-ionic surfactants.
 - 4. The composition of claim 1, wherein the fragrance composition comprises
- a) at least 10.0% by weight of a fragrance composition of one or more perfume ingredients having a boiling point of 250 °C or less and ClogP of less than 3.0;
- b) at least 15% by weight of a fragrance composition of one or more perfume ingredients having a boiling point of 250 °C or less and ClogP of 3.0 or more; and
- c) at least 20% by weight of a fragrance composition of one or more perfume ingredients have a boiling point of greater than 250 °C.
 - 5. The composition of claim 4, wherein the fragrance composition comprises:
- a) at least 20.0% by weight of the fragrance composition of one or more perfume ingredients having a boiling point of 250 °C or less and ClogP of less than 3.0;

b) at least 15% by weight of the fragrance composition of one or more perfume ingredients having a boiling point of 250 °C or less and ClogP of 3.0 or more; and

- c) at least 30% by weight of the fragrance composition of one or more perfume ingredients have a boiling point of greater than 250 °C.
 - 6. The composition of claim 5, wherein the fragrance composition comprises:
 - a) 30.0 to 40.0 % by weight of the fragrance composition of one or more perfume ingredients having a boiling point of 250 °C or less and ClogP of less than 3.0;
 - b) 20.0 to 25.0 % by weight of the fragrance composition of a one or more perfume ingredients having a boiling point of 250 °C or less and ClogP of 3.0 or more; and
 - c) 40.0 to 45.0 % by weight of the fragrance composition of one or more perfume ingredients have a boiling point of greater than 250 °C.
- 7. The composition of claim 1, wherein the silicone: fragrance composition weight ratio is from 1:1 to 10:1.
 - 8. A composition comprising:

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- i) a fragrance composition, wherein the fragrance composition comprises
- a) at least 10.0% by weight of a fragrance composition of one or more perfume ingredients having a boiling point of 250 °C or less and ClogP of less than 3.0,
 - b) at least 20% by weight of a fragrance composition of one or more perfume ingredients having a boiling point of 250 °C or less and ClogP of 3.0 or more, and
 - c) at least 30% by weight of a fragrance composition of one or more perfume ingredients have a boiling point of greater than 250 °C; and
 - ii) a silicone-in-water emulsion comprising at least one surfactant.
- 9. The composition of claim 8, wherein the dispersed silicone has a viscosity of
 100 million mm²s⁻¹ or more and a particle size of 2 μm or less.

10. The composition of claim 9, wherein the silicone-in-water emulsion is non-ionic and the silicone is a linear polysiloxane polysilalkylene copolymer, wherein the monovalent silicone substituents are substantially all C_{1-6} hydrocarbon groups and the copolymer has a viscosity of 150 million mm²s⁻¹ or more and a particle size of 1 μ m or less.

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- 11. The composition of claim 10, wherein the silicone-in-water emulsion is a non-ionic emulsion comprising about 60 % by weight of a linear silicone copolymer formed by the hydrosilylation of a Si-H end-blocked polydimethyl siloxane and a vinylend-blocked polydimethyl siloxane, wherein the copolymer has a viscosity of about 170 mm²s⁻¹ and a particle size of about 0.5 μ m, and about 5 % by weight of non-ionic surfactants.
 - 12. The composition of claim 8, wherein the fragrance composition comprises
- a) at least 20.0% by weight of the fragrance composition of one or more perfume ingredients having a boiling point of 250 °C or less and ClogP of less than 3.0;
- b) at least 20% by weight of the fragrance composition of one or more perfume ingredients having a boiling point of 250 °C or less and ClogP of 3.0 or more; and
- c) at least 30% by weight of the fragrance composition of one or more perfume ingredients have a boiling point of greater than 250 °C.
- 13. The composition of claim 12, wherein the fragrance composition comprises:
- a) 30.0 to 40.0 % by weight of the fragrance composition of one or more perfume ingredients having a boiling point of 250 °C or less and ClogP of less than 3.0;
- 25 b) 20.0 to 25.0 % by weight of the fragrance composition of a one or more perfume ingredients having a boiling point of 250 °C or less and ClogP of 3.0 or more; and
 - c) 40.0 to 45.0 % by weight of the fragrance composition of one or more perfume ingredients have a boiling point of greater than 250 $^{\circ}$ C.
- 14. The composition of claim 8, wherein the silicone: fragrance eomposition weight ratio is from 1:1 to 10:1.

15. A composition, wherein the composition is a hair care preparation, comprising:

- i) a fragrance composition; and
- ii) a silicone-in-water emulsion comprising at least one surfactant,
 5 wherein the dispersed silicone has a viscosity of 100 million mm²s⁻¹ or more and a particle size of less than 2 μm.
 - 16. The hair care preparation of claim 15, wherein the silicone: fragrance composition weight ratio is from 1:1 to 10:1.
- 17. The hair care preparation of claim 15, wherein the hair care preparation is a shampoo, a conditioner, or a conditioning shampoo.
 - 18. A composition, wherein the composition is a hair care preparation, comprising:
 - i) a fragrance composition, wherein the fragrance composition comprises
 - a) at least 10.0% by weight of a fragrance composition of one or more perfume ingredients having a boiling point of 250 °C or less and ClogP of less than 3.0;
 - b) at least 20% by weight of a fragrance composition of one or more perfume ingredients having a boiling point of 250 °C or less and ClogP of 3.0 or more; and
 - c) at least 30% by weight of a fragrance composition of one or more perfume ingredients have a boiling point of greater than 250 °C; and
 - ii) a silicone-in-water emulsion comprising at least one surfactant.
- 19. The hair care preparation of claim 18, wherein the silicone: fragrance composition weight ratio is from 1:1 to 10:1.
 - 20. The hair care preparation of claim 18, wherein the hair care preparation is a shampoo, a conditioner, or a conditioning shampoo.
 - 21. A method of cleansing hair, which comprises applying to the hair a hair care preparation comprising:
- i) a fragrance composition; and

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ii) a silicone-in-water emulsion comprising at least one surfactant, wherein the dispersed silicone has a viscosity of 100 million $mm^2s^{\text{-1}}$ or more and a particle size of less than 2 μm .

- 22. The method of claim 21, wherein the silicone: fragrance composition weight ratio is from 1:1 to 10:1.
 - 23. The method of claim 21, wherein the hair care preparation is a shampoo, a conditioner, or a conditioning shampoo.
 - 24. A method of cleansing hair, which comprises applying to the hair a hair care preparation comprising:
 - i) a fragrance composition, wherein the fragrance composition comprises

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- a) at least 10.0% by weight of a fragrance composition of one or more perfume ingredients having a boiling point of 250 °C or less and ClogP of less than 3.0;
- b) at least 20% by weight of a fragrance composition of one or more perfume ingredients having a boiling point of 250 °C or less and ClogP of 3.0 or more; and
 - c) at least 30% by weight of a fragrance composition of one or more perfume ingredients have a boiling point of greater than 250 $^{\circ}$ C; and
 - ii) a silicone-in-water emulsion comprising at least one surfactant.
 - 25. The method of claim 24, wherein the silicone: fragrance composition weight ratio is from 1:1 to 10:1.
 - 26. The method of claim 24, wherein the hair care preparation is a shampoo, a conditioner, or a conditioning shampoo.